

Synthetic Strategy and Pharmacological Approaches of Benzotriazole; A-Review

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ABSTRACT: This review attempts to bring tighter various developments in synthesis and biological activities of benzotriazole derivatives. Benzotriazole is bicyclic heterocyclic compound contain three nitrogen atom fused in benzene ring. This has wide range of biological and pharmacological activity. Literature study reveals that Nitrogen containing heterocyclic compound reported various biological effect specially benzotriazole which is containing three nitrogens at 1,3,4 position which is essential for synthesis of novel molecules.

KEYWORDS: Benzotriazole, Biological Activity, synthesis, pharmacological activity.

I. INTRODUCTION

BENZOATRIAZOLE

In late 1960, azoles are a heterocyclic compound class that constitutes the largest group of available antifungal drugs. Particularly, the imidazole ring is the chemical component that confers activity to azoles. Triazoles are obtained by a slight modification of this ring and similar or improved activities as well as fewer adverse effects are reported for triazole derivatives. Consequently, it is not surprising that benzimidazole/benzotriazole derivatives are biologically active¹.

Benzo-fused azoles are a class of heterocyclic compounds of great interest in the pharmaceutical chemistry area. Benzimidazole and its derivatives have been studied for decades, and drugs having this heterocycle moiety as a main element have been widely used in clinics, for instance as anthelmintic in humans^{2,3}. Benzo-condensed azole containing three heteroatom's, such as bezoxadiazole, benzothiazole, and benzotriazole has been extensively studied for their broad range of biological activity^{4,5}.

The primary triazole molecule is comprised of a five-member heterocyclic compound with three nitrogen atoms and a vicinal pair of carbon atoms.

Two isomers are presented, with the hydrogen present in either the 1-H (Figure 2(a)) or 2-H configuration (Figure 2(b)). Benzotriazole is formed when a benzene ring is fused to the triazole (e.g., Figure 1(a)). Many Benzotriazole compounds exist, but those with a high production volume (HPV) and which have been reported as being present in the environment most commonly follow these two configurations.

Although a significant numbers of engine valve-actuation systems including cam-based and cam less mechanisms have been already introduced by several researches and companies, only few types of these systems (mainly cam-based) have been employed on commercial vehicles due to the liability, durability and cost issues. Cam-based valve systems offer reliable and durable functionality, the cam less valve trains can vary valve lift and more timing to a greater extent comparing to the cam-based types. Among various categories of cam less mechanisms, the electromagnetic actuator system is the most desired one.

Benzotriazoles used as anticorrosive additives have hydrogen oriented in the 1-H position (Figure 1(a)). In addition, two BZTs used extensively for this purpose are methylated, with the functional group at either the 4' or 5' positions (e.g., 4-TT and 5-TT) (Figure 1(b-d)). Conversely, the UV-stabilizing BZTs (UV-BZTs) generally have a phenolic moiety positioned at 2-H (e.g., Figure 2e). Overall, the phenolic BZTs are higher in molecular weight (MW) and structurally more complex. These compounds often include an ester, chlorine atom, or additional functional groups (e.g., phenol and methyl) as part of the molecule.

Stability is a key characteristic and desired attribute of most commercially used Benzotriazole. The parent structure will in many instances readily undergo a range of reactions; however, the triazole moiety in most cases remains unchanged⁶. For applications utilizing their anticorrosive properties, Benzotriazoles form highly stable metal complexes

with numerous metals and alloys, particularly copper, which protect them against corrosive attack. In this application, the Benzotriazole-metal complex forms a passive layer on the metal surface, preventing surface reactions⁷.

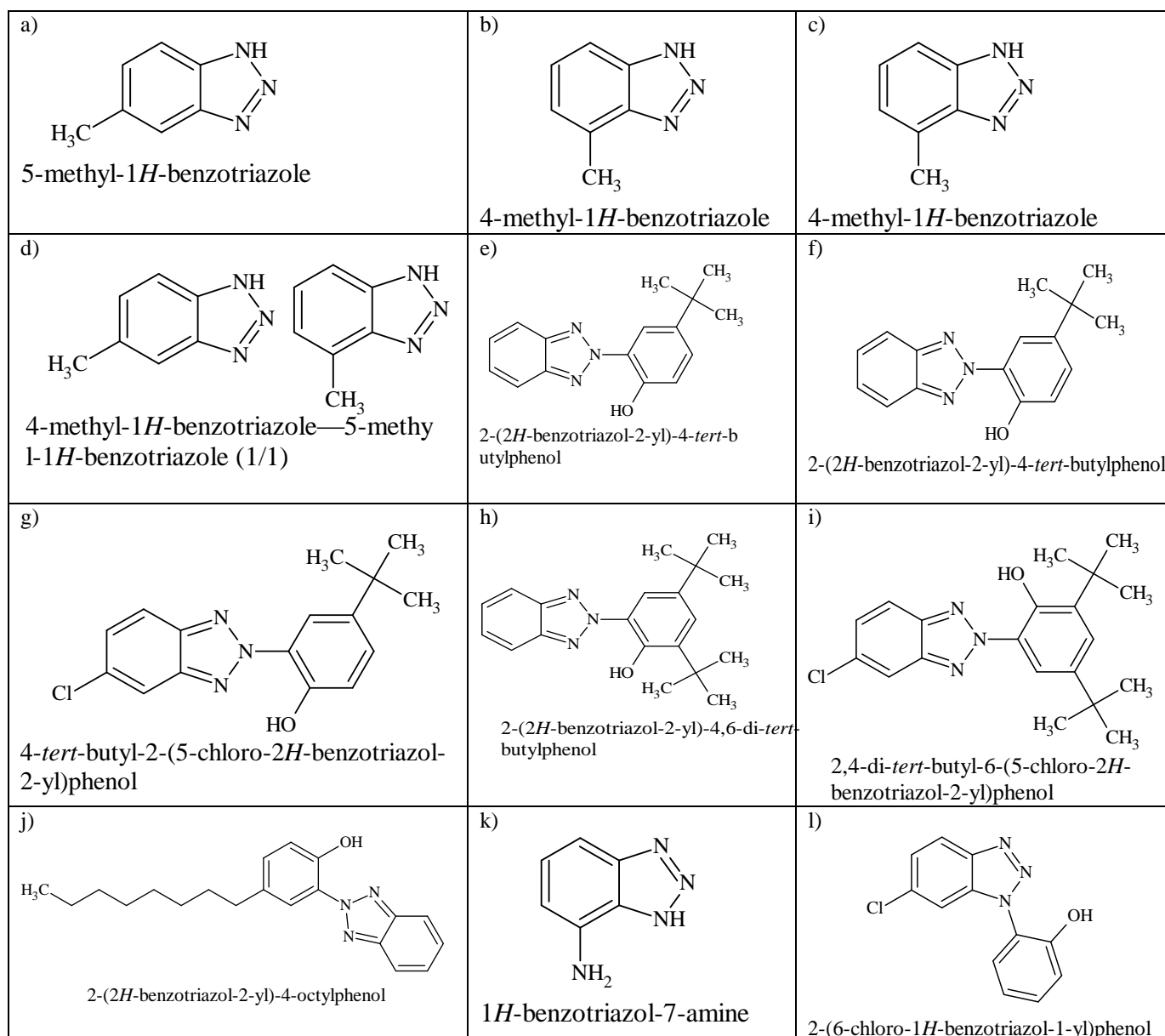


FIGURE 1 Structures, names of select Benzotriazole compounds

Structure of benzotriazole

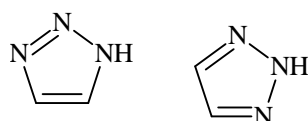
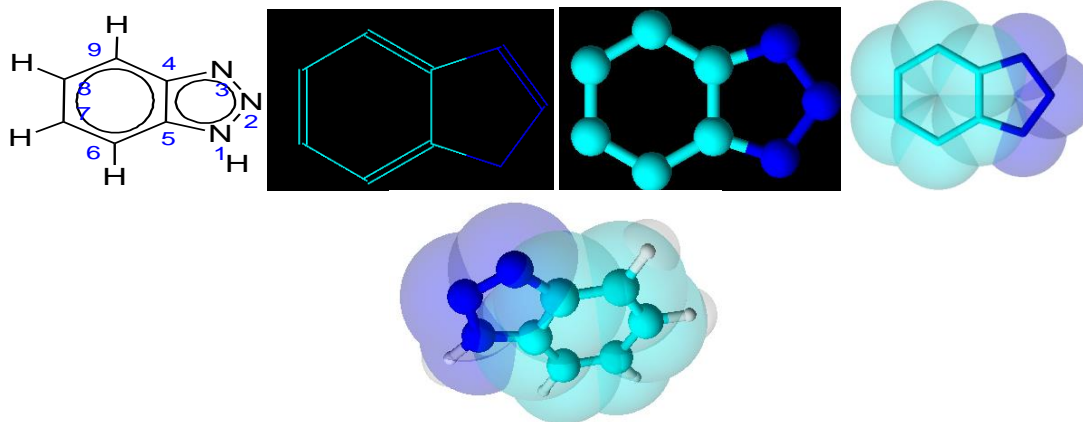


FIGURE 2 Structures of 1-H (a) and 2-H (b) triazole.



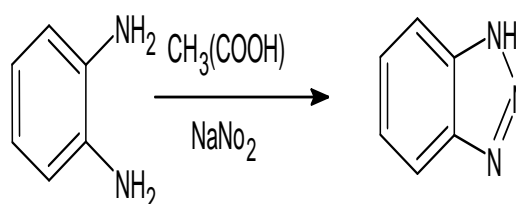
Physicochemical properties

Molecular Formula:	H2N2
Formula Weight:	30.02928
Composition:	H(6.71%) N(93.29%)
Molar Refractivity:	6.36 ± 0.5 cm ³
Molar Volume:	24.3 ± 7.0 cm ³
Parachor:	68.6 ± 8.0 cm ³
Index of Refraction:	1.435 ± 0.05
Surface Tension:	63.0 ± 7.0 dyne/cm
Density:	1.23 ± 0.1 g/cm ³
Dielectric Constant:	Not available
Polarizability:	2.52 ± 0.5 10 ⁻²⁴ cm ³
RDBE:	1
Monoisotopic Mass:	30.021798 Da
Nominal Mass:	30 Da
Average Mass:	30.0293 Da
M+:	30.021249 Da
M-:	30.022347 Da
[M+H]+:	31.029075 Da
[M+H]-:	31.030172 Da
[M-H]+:	29.013424 Da
[M-H]-:	29.014522 Da

Chemistry and synthesis of Benzotriazole-

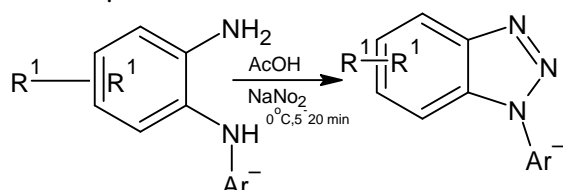
Benzotriazole display a broad range of biological activities and are found in many potent, biologically active compounds. So far, modifications of the benzotriazole ring have proven highly effective with improved potency and lesser toxicity. Benzotriazole also has three tautomers. Namely two 1H-forms and one 2H-form. In solution, the equilibrium lies almost entirely on the side of the 1H-forms ⁸.

It is synthesized by diazotization process using benzene-1,2-diamine with sodium nitrite and acetic acid.



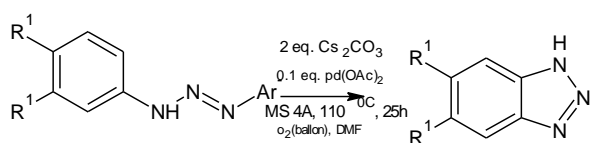
Cyclocondensation of (arylamino)aryliminophosphoranes : (9)

A cyclocondensation of 2-(arylamino)aryliminophosphoranes enables the synthesis of 1-aryl-1,2,3-benzotriazoles under mild conditions. The reaction involves a three-step, Halogen-free route starting from simple nitroarenes and aryl amines



C-H activation of aryl triazene: (10)

C-H activation of aryl triazene compounds followed by intermolecular amination in the presence of a catalytic amount of Pd(OAc)₂ provides 1-aryl-1H-benzotriazoles at moderate temperature



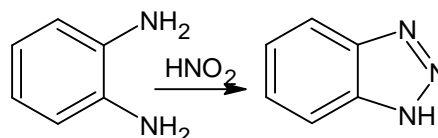
1, 7-palladium migration-cyclization-dealkylation(11)

A 1, 7-palladium migration-cyclization-dealkylation sequence for the regioselective synthesis of benzotriazoles proceed in excellent yields with high regioselectivities



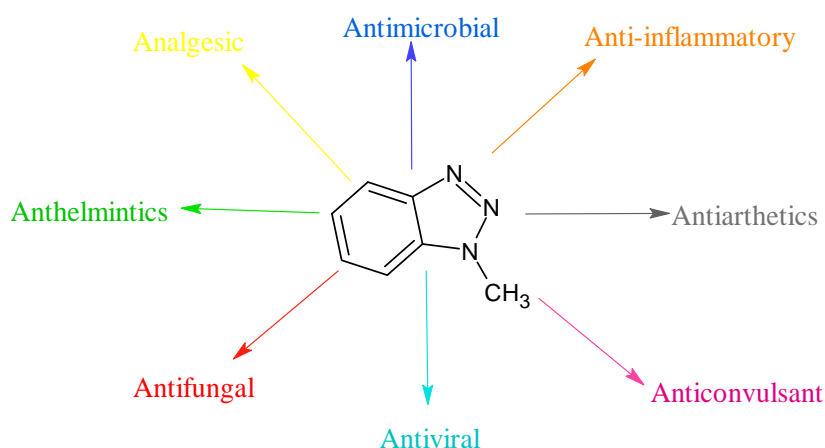
Cyclocondensation of o-phenylenediamines (12)

The standard synthesis of benzotriazoles is the cyclocondensation of o-phenylenediamines with sodium nitrite in acetic acid.



Biological activity-

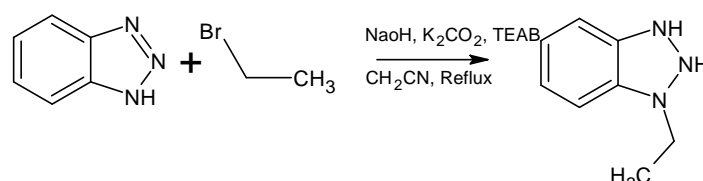
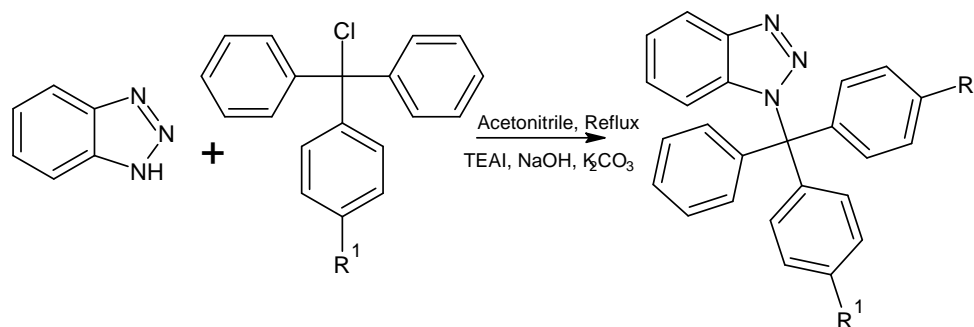
Synthetic compounds of derivatives of Benzotriazole have shown analgesic, antibacterial, antifungal activities¹⁻⁶ ant filarial activities, and Benzotriazole also reported for anticonvulsant and anti-inflammatory, antitumor activities; literature study reveals the antiviral activity¹³



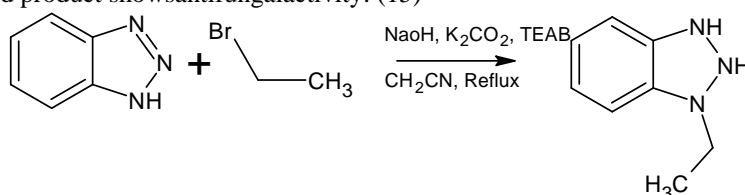
Antifungal activity

Rezaei Z.et. al was reported Design, synthesis, and antifungal activity of triazole and benzotriazole derivatives. In this study a series of new 1,2,4-triazole and benzotriazoles related to clotrimazole and fluconazole were synthesized and then the compounds docked in the active site of 14a-demethylase using Autodock program. In

addition we investigated the activity of the compounds against *Microsporumcanis*, *Trichophyton mentagrophyte*, *Trichophyton rubrum*, *Epidermophyton floccosum*, and *Candidaalbicans*.(14)

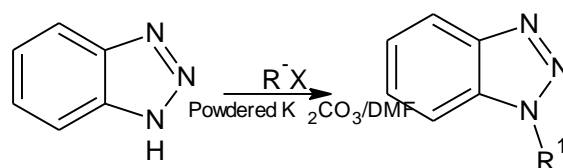


S. Khabnadideh et. al were reported Synthesis and antifungal activity of benzimidazole, benzotriazole and aminothiazole derivatives the compound synthesized by the reaction of benzotriazole with bromo-alkanesth synthesized product shows antifungal activity. (15)

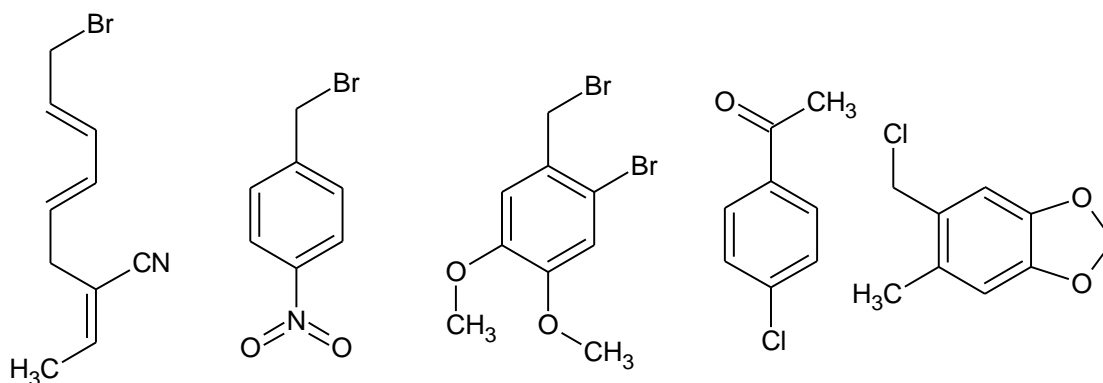


Antimicrobial activity

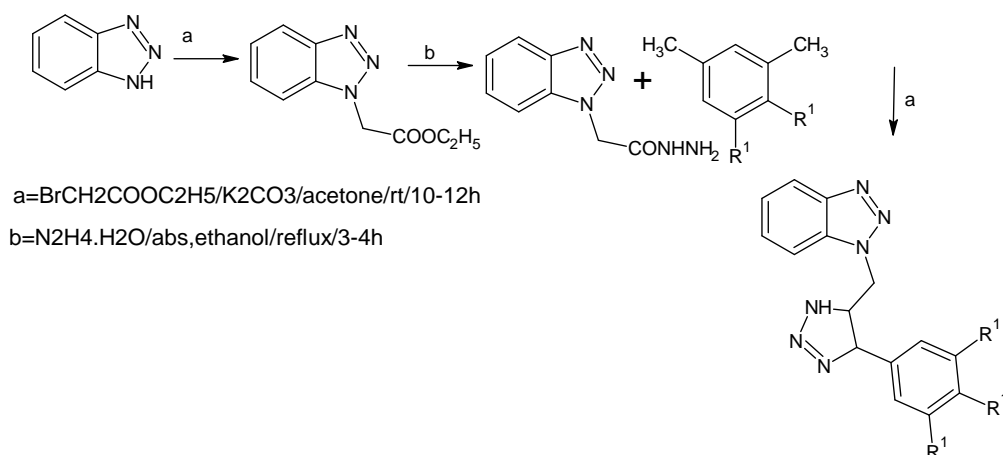
S. Nanjunda Swamy et.al was reported Microwave-assisted synthesis of N-alkylated benzotriazole derivatives: Antimicrobial studies.(16)



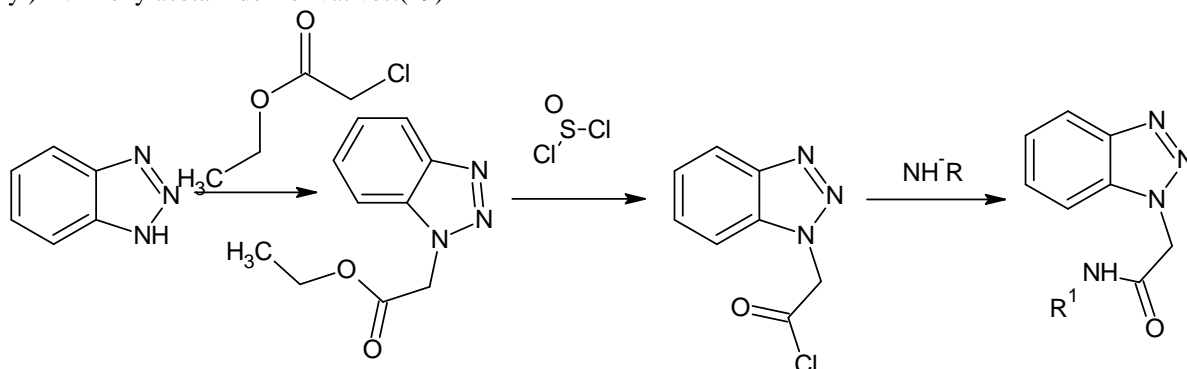
Where: R-X=



Mahesh chand et .al were reported Synthesis and antimicrobial and antioxidant activities of hybrid molecules containing benzotriazole and 1,2,4-triazole.(17-18)

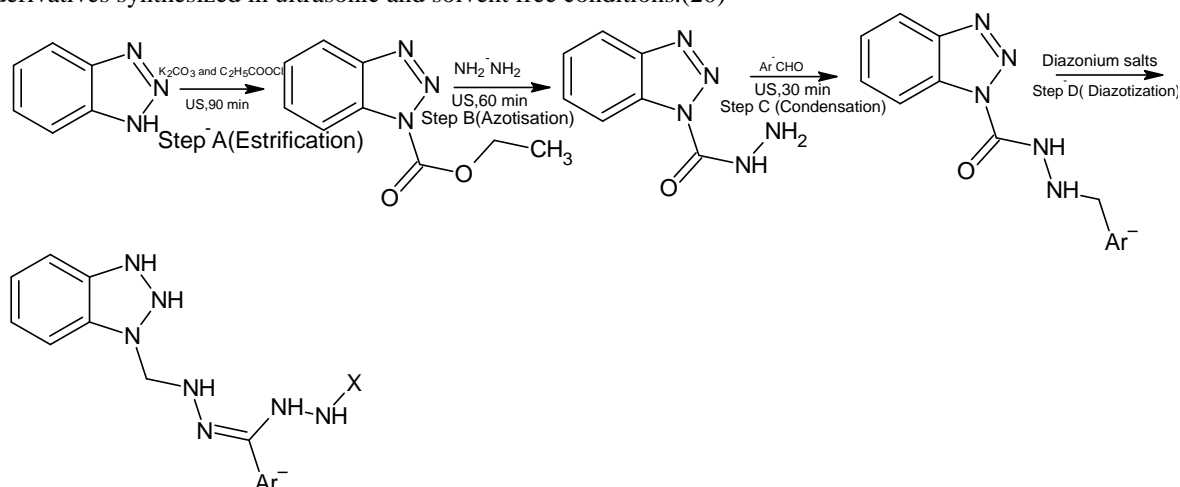


C. M. Jamkhandi et al were reported Synthesis and Antimicrobial Evaluation of 2-(1H-1,2,3-Benzotriazol-1-yl)- N-Phenylacetamide Derivatives.(19)



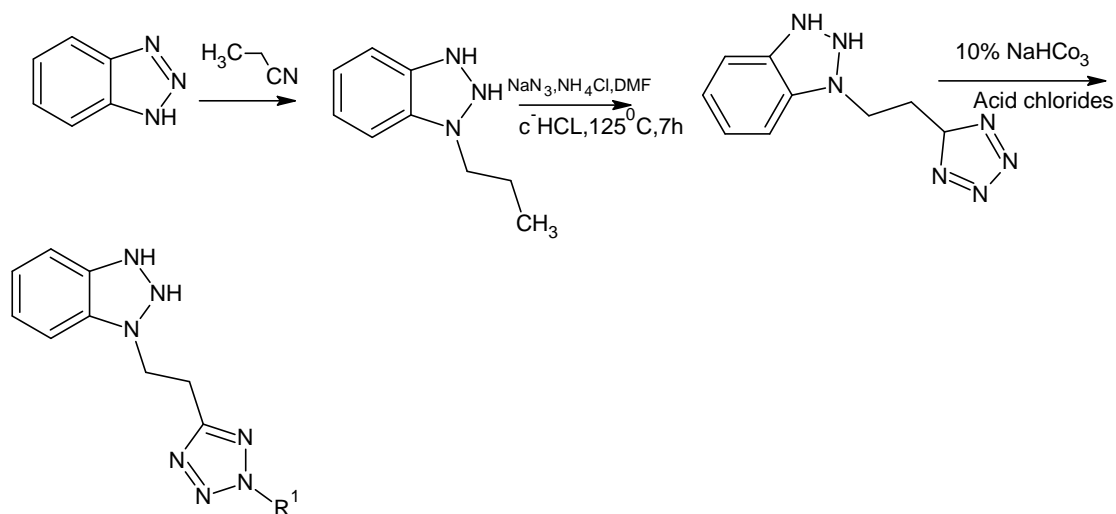
Anthelmintic activity

M.S Sudhir et.al were reported Evaluation of in vitro anthelmintic activities of novel 1,2,3 e benzotriazole derivatives synthesized in ultrasonic and solvent free conditions.(20)



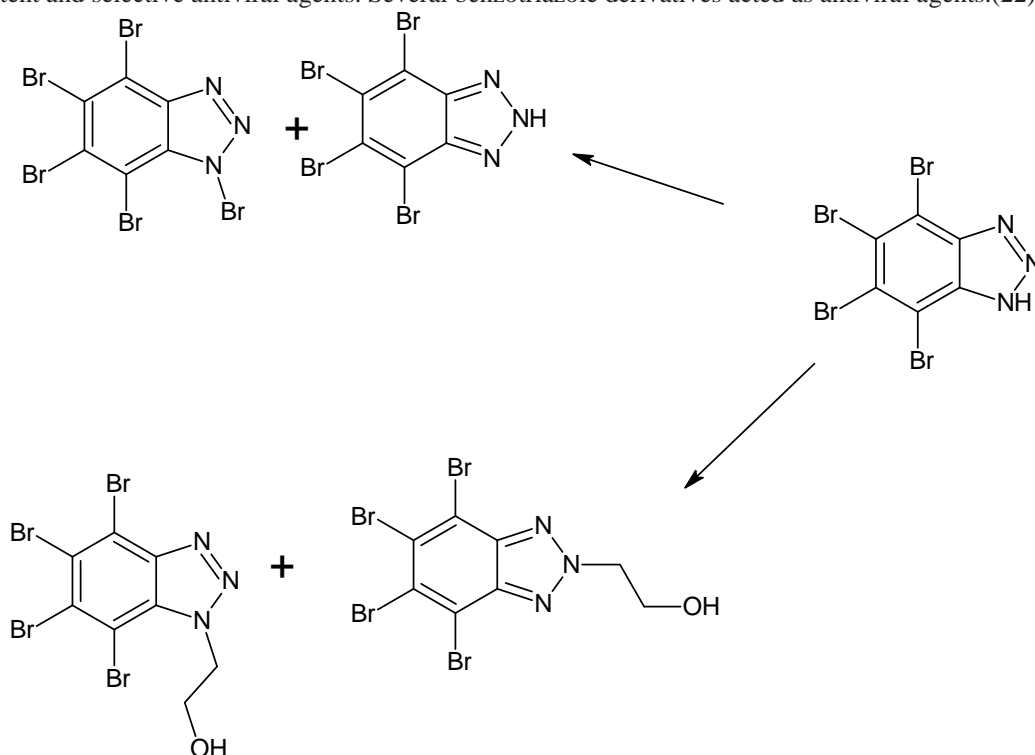
Anti-inflammatory Activity

aiyalurajasekaranet al were reported Synthesis of some novel triazole derivatives as anti-nociceptive and anti-inflammatory agents.(21)



Antiviral agent

Briguglio et al were reported versatile biological behaviour of benzotriazole. Nucleosides and non-nucleosides inhibitors are widely investigated to implement the pharmacological arsenal and to obtain more potent and selective antiviral agents. Several benzotriazole derivatives acted as antiviral agents.(22)



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